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10/572,674	08/30/2006	Takashi Fujimaki	Q93741	4433
23373 SUGHRUE MI	7590 10/01/200 ON. PLLC	EXAMINER		
2100 PENNSYLVANIA AVENUE, N.W.			YI, STELLA KIM	
SUITE 800 WASHINGTON, DC 20037			ART UNIT	PAPER NUMBER
			1791	
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Please find below and/or attached an Office communication concerning this application or proceeding.

The time period for reply, if any, is set in the attached communication.

	Application No.	Applicant(s)
	10/572,674	FUJIMAKI ET AL.
Office Action Summary	Examiner	Art Unit
	Stella Yi	1791
The MAILING DATE of this communication app Period for Reply	ears on the cover sheet with the c	orrespondence address
A SHORTENED STATUTORY PERIOD FOR REPLY WHICHEVER IS LONGER, FROM THE MAILING DA - Extensions of time may be available under the provisions of 37 CFR 1.13 after SIX (6) MONTHS from the mailing date of this communication. - If NO period for reply is specified above, the maximum statutory period w - Failure to reply within the set or extended period for reply will, by statute, Any reply received by the Office later than three months after the mailing earned patent term adjustment. See 37 CFR 1.704(b).	ATE OF THIS COMMUNICATION 36(a). In no event, however, may a reply be tim vill apply and will expire SIX (6) MONTHS from cause the application to become ABANDONE	N. nely filed the mailing date of this communication. D (35 U.S.C. § 133).
Status		
Responsive to communication(s) filed on <u>30 Au</u> This action is FINAL . 2b)☑ This Since this application is in condition for allowar closed in accordance with the practice under E	action is non-final. nce except for formal matters, pro	
Disposition of Claims		
4) ☐ Claim(s) 1-11 is/are pending in the application. 4a) Of the above claim(s) is/are withdraw 5) ☐ Claim(s) is/are allowed. 6) ☐ Claim(s) 1-11 is/are rejected. 7) ☐ Claim(s) is/are objected to. 8) ☐ Claim(s) are subject to restriction and/or Application Papers 9) ☐ The specification is objected to by the Examine 10) ☐ The drawing(s) filed on is/are: a) ☐ access	relection requirement. r. epted or b)□ objected to by the B	
Applicant may not request that any objection to the o	on is required if the drawing(s) is obj	ected to. See 37 CFR 1.121(d).
11) The oath or declaration is objected to by the Ex	aminer. Note the attached Office	Action or form PTO-152.
Priority under 35 U.S.C. § 119		
 12) Acknowledgment is made of a claim for foreign a) All b) Some * c) None of: 1. Certified copies of the priority documents 2. Certified copies of the priority documents 3. Copies of the certified copies of the prior application from the International Bureau * See the attached detailed Office action for a list of the certified copies 	s have been received. s have been received in Applicati ity documents have been receive ı (PCT Rule 17.2(a)).	on No ed in this National Stage
Attachment(s) 1) Notice of References Cited (PTO-892) 2) Notice of Draftsperson's Patent Drawing Review (PTO-948) 3) Information Disclosure Statement(s) (PTO/SB/08) Paper No(s)/Mail Date 03/20/2006.	4) Interview Summary Paper No(s)/Mail Da 5) Notice of Informal P 6) Other:	ate

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DETAILED ACTION

Claim Rejections - 35 USC § 103

- 1. The following is a quotation of 35 U.S.C. 103(a) which forms the basis for all obviousness rejections set forth in this Office action:
 - (a) A patent may not be obtained though the invention is not identically disclosed or described as set forth in section 102 of this title, if the differences between the subject matter sought to be patented and the prior art are such that the subject matter as a whole would have been obvious at the time the invention was made to a person having ordinary skill in the art to which said subject matter pertains. Patentability shall not be negatived by the manner in which the invention was made.
- 2. Claims 1-4 and 7-11 are rejected under 35 U.S.C. 103(a) as being unpatentable over ABELLA (3,903,294) and in further view of KOBAYASHI et al. (JP 2003-251673).

Regarding claims 1-3 and 7, ABELLA teach a process of forming heat shrinkable polyethylene terephalate packaging film (PET) wherein heating of the polymerization reaction of the PET melt is conducted at 250°C; then extruded through a die onto a cooled casting rolls to from a cast sheet of PET; and thereafter biaxially orienting the amorphous cast sheet by stretching the sheet in a first direction followed by stretching in a second direction perpendicular to the first direction (Col.4, lines 2-20). ABELLA teach that the said PET melt is produced by transesterification reaction of a dialkyl ester terephthalic with at least two molecular proportions of ethylene glycol per molecular proportion of dialkyl terephthalate and that many known catalysts can be used to speed the transesterification reaction (Col.3, lines 12-31). Furthermore, ABELLA teach that PET has an intrinsic viscosity of about 0.62 to 0.9 deciliter/gram (Col.3, line 59). ABELLA does not explicitly disclose the said PET comprising the specific mixture "A" of instant claim 1. However, KOBAYASHI et al. discloses a method for manufacturing heat-resistant sheet and molded body of polyethylene terephthalate polyester (PET)

comprising melting at a temperature about 260-290°C (Page 12, [0025]) of an ingredient (mixture) A comprising:

- (1) 100 parts by weight of a polyethylene terephthalate (PET)-based polyester "a" having a melt flow rate (MFR, JIS method: 280, load 2.16kg) of 45-130g/10minutes as a main raw material (Page 5, [0009] and Page 7, [0010]);
- (2) 10 to 100 parts by weight of ethylene glycol/cyclohexane dimethanol/phthalic acid copolyester (Page 10, [0017]);
 - (3) 20 parts by weight of PBT (polyester elastomer) (Page 16, [0036]);
- (4) 0 to 100 parts by weight a mixture containing a compound "b" ("d") having two epoxy groups and a compound "c" ("e") having three or more epoxy groups in a weight ratio of 95/5 as a binder (Page 9, [0016]); and
- (5) 0.25 part by weight of an organic acid metal salt "g" as a catalyst (Page 5, [0009]);

turning the mixture "A" into block copolymer pellets by subjecting it to a uniform reaction under dearation and dehydration in vacuum (Page 5, [0009] and Page 15, [0032]); and

molding a mixture comprising 100 parts by weight of the block copolymer pellets (Page 15, [0032]).

It would have been obvious to one of ordinary skill in the art to have substituted the PET of KOBAYASHI et al. for the PET of ABELLA for the predictable results of forming a PET based block copolymer heat-shrinkable packaging film. In addition, ABELLA teach that a PET film having a balanced heat shrinkable property along with

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high heat resistance is desired (Col.1, lines 32-40). The PET film of KOBAYASHI et al. is manufactured to be a high heat-resistant film and is manufactured by an extrusion method such as the one taught by ABELLA. Therefore, it would have been obvious to one of ordinary skill in the art to have modified the method of forming a heat shrinkable PET of ABELLA by incorporating the mixture A of KOBAYASHI et al. in order to form a PET-based copolymer packaging film with balanced heat shrinkable property and high heat resistance.

Regarding claim 4, ABELLA teach that a temperature for extending the film into the oriented film through a biaxial orientation method is 65 to 85°C (Col.4, lines 34-45).

Regarding claim 8, KOBAYASHI et al. teach that compound "b" ("d") comprises at least one selected from the group consisting of ethylene glycol diglycidyl ether of an aliphatic series system, polyethylene glycol diglycidyl ether, alicyclic system hydrogenation bisphenol A diglycidyl ether (Page 6, [0009]).

Regarding claim 9, KOBAYASHI et al. teach that compound "c" ("e") comprises at least one selected from the group consisting of trimethylolpropane triglycidyl ether of an aliphatic series system, glycerol triglycidyl ether, triglycidyl paller of triglycidyl isocyanurate of a heterocyclic system, epoxidized soybean oil, epoxidation linseed oil, phenol novalak epoxy resin, and cresolnovolak epoxy resin (Page 6, [0009] and Page8, [0015]).

Regarding claim 10, KOBAYASHI et al. teach that the coupling reaction catalyst "g" comprises a composite containing salts selected from lithium salt, stearic acid, acetic

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acid, sodium salt, potassium salt, magnesium salt, calcium salt, and zinc salt (Page 6, [0009]).

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Regarding claim 11, ABELLA teach a process of forming heat shrinkable polyethylene terephalate packaging film (PET) wherein heating of the polymerization reaction of the PET melt is conducted at 250°C; then extruded through a die onto a cooled casting rolls to from a cast sheet of PET; and thereafter biaxially orienting the amorphous cast sheet by stretching the sheet in a first direction followed by stretching in a second direction perpendicular to the first direction (Col.4, lines 2-20). ABELLA teach that the said PET melt is produced by transesterification reaction of a dialkyl ester terephthalic with at least two molecular proportions of ethylene glycol per molecular proportion of dialkyl terephthalate and that many known catalysts can be used to speed the transesterification reaction (Col.3, lines 12-31). Furthermore, ABELLA teach that PET has an intrinsic viscosity of about 0.62 to 0.9 deciliter/gram (Col.3, line 59). ABELLA does not explicitly disclose the said PET comprising the specific mixture "A" of instant claim 1. However, KOBAYASHI et al. discloses a method for manufacturing heat-resistant sheet and molded body of polyethylene terephthalate polyester (PET) comprising melting at a temperature about 260-290°C (Page 12, [0025]) of an ingredient (mixture) A comprising:

(1) 100 parts by weight of a polyethylene terephthalate (PET)-based polyester "a" having a melt flow rate (MFR, JIS method: 280, load 2.16kg) of 45-130g/10minutes as a main raw material (Page 5, [0009] and Page 7, [0010]);

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- (2) 10 to 100 parts by weight of ethylene glycol/cyclohexane dimethanol/phthali acid copolyester (Page 10, [0017]);
 - (3) 20 parts by weight of PBT (polyester elastomer) (Page 16, [0036]);
- (4) 1 to 15 parts by weight of a binder masterbatch comprising 50 to 100 parts by weight of a mixture containing a compound "b" ("d") having two epoxy groups and a compound "c" ("e") having three or more epoxy groups in a weight ratio of 95/5 as a binder (Page 9, [0016]) and 100 parts by weight of a base substance "h" as a binder (Page 11, [0021]); and
- (5) 0.25 to 10 parts by weight of a catalyst masterbatch containing 5 or less parts by weight of a catalyst "g" and 100 parts by weight of a base substance as a catalyst (Page 11, [0021]-[0022]);

turning the mixture "A" into block copolymer pellets by subjecting it to a uniform reaction under dearation and dehydration in vacuum (Page 11, [0022]-[0026]);

molding a mixture comprising 100 parts by weight of the block copolymer pellets (Page 11, [0022]-[0026]).

It would have been obvious to one of ordinary skill in the art to have substituted the PET of KOBAYASHI et al. for the PET of ABELLA for the predictable results of forming a PET based block copolymer heat-shrinkable packaging film. In addition, ABELLA teach that a PET film having a balanced heat shrinkable property along with high heat resistance is desired (Col.1, lines 32-40). The PET film of KOBAYASHI et al. is manufactured to be a high heat-resistant film and is manufactured by an extrusion method such as the one taught by ABELLA. Therefore, it would have been obvious to

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one of ordinary skill in the art to have modified the method of forming a heat shrinkable PET of ABELLA by incorporating the mixture A of KOBAYASHI et al. in order to form a PET-based copolymer packaging film with balanced heat shrinkable property and high heat resistance.

3. Claims 5-6 are rejected under 35 U.S.C. 103(a) as being unpatentable over ABELLA (3,903,294) in view of KOBAYASHI et al. (JP 2003-251673) as applied to claims 1-4 and 7-11 above and in further view of ERICKSON (3,631,899).

The teachings of ABELLA and KOBAYASHI et al. are applied as described above for claims 1-4 and 7-11.

Regarding claim 5, ABELLA teach a process of forming heat shrinkable polyethylene terephalate packaging film (PET) by biaxially orienting the amorphous cast sheet by stretching the sheet in a first direction followed by stretching in a second direction perpendicular to the first direction (Col.4, lines 2-20). ABELLA teach that the said PET film is biaxially oriented by balanced shrinkage properties of at least 40 percent in both longitudinal and transverse directions (Col.2, lines 54-57) but is silent to the temperature of the heat shrinkage being at 130°C. ABELLA teach that the greater the shrinkage the more useful is the film as a heat shrinkable film. In other words, film having higher shrinkage characteristics results in a more useful film (Col.6, lines 55-70) However, ERICKSON teach a PET film that shrinks between about 40 and 50 percent at 130°C and that the film should not be heated to a temperature above 130°C to assure that the film will exhibit the desired heat-shrinkage characteristics at practical shrinking

temperatures (Col.3, lines 38-57). Therefore, it would have been obvious to one of ordinary skill in the art that the PET film of ABELLA would have a degree of heat-shrinkage of 30% or more at 130°C.

Regarding claim 6, modified ABELLA is silent to a weld-cut sealing strength. However, ABELLA modified by KOBAYASHI et al. teach a method of producing a heat-shrinkable film of a PET based block copolymer polyester comprising the formula of mixture A of instant claim 1 and it would have been obvious to one of ordinary skill in the art that the weld-cut sealing strength would depend on the composition of the PET film mixture. Therefore, it would have been obvious to one of ordinary skill in the art that the PET film of KOBAYASHI et al. would have strength of 500g/15mm width or more since it comprises the ingredients as that of mixture A of instant claim 1.

Conclusion

Any inquiry concerning this communication or earlier communications from the examiner should be directed to Stella Yi whose telephone number is 571-270-5123. The examiner can normally be reached on Monday - Thursday from 8:00 AM to 5:00 PM.

If attempts to reach the examiner by telephone are unsuccessful, the examiner's supervisor, Christina Johnson can be reached on 571-272-1176. The fax phone number for the organization where this application or proceeding is assigned is 571-273-8300.

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Information regarding the status of an application may be obtained from the Patent Application Information Retrieval (PAIR) system. Status information for published applications may be obtained from either Private PAIR or Public PAIR. Status information for unpublished applications is available through Private PAIR only. For more information about the PAIR system, see http://pair-direct.uspto.gov. Should you have questions on access to the Private PAIR system, contact the Electronic Business Center (EBC) at 866-217-9197 (toll-free). If you would like assistance from a USPTO Customer Service Representative or access to the automated information system, call 800-786-9199 (IN USA OR CANADA) or 571-272-1000.

SY

/Matthew J. Daniels/ Primary Examiner, Art Unit 1791 9/11/09